

Title: Lithium facts periodic table

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What atomic number is lithium?

Lithium is atomic number 3 on the periodic table with element symbol Li. Here are a collection of lithium facts, including its properties, uses, and sources. Basic Lithium Facts Name: Lithium Atomic Number: 3 Element Symbol: Li Group: 1 Period: 2 Block: s Element Family: Alkali Metal Electron Configuration: [He]2s 1 (shorthand) or 1s 2 2s 1 (full)

Where is lithium located on the periodic table?

Lithium is a chemical element with the symbol Li and atomic number 3. Lithium is the 3 element on the periodic table. It is located in group 1 and period 2 in the modern periodic table. 5. How do you find Lithium on the periodic table? Lithium is the 3 element on the periodic table.

What group does lithium belong to?

It is a soft, silver-white metal belonging to the Alkali Metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Lithium belongs to group 1 of the periodic table having trivial name alkali metals*.

What do we know about lithium?

What we know about lithium: Lithium is the third element in the periodic table, with three protons and the element symbol Li. It has an atomic mass of 6.941. Natural lithium is a mixture of two stable isotopes, lithium-6 and lithium-7. Lithium-7 accounts for over 92% of the natural abundance of the element. Lithium is an alkali metal.

Lithium is a reactive element of the periodic table, Learn interesting facts about lithium, its discovery, uses, compounds and more

Lithium was discovered from a mineral, while other common alkali metals were discovered from plant material. This is thought to explain the origin of the element's name; from "lithos" (Greek for "stone"). ...

Here are 10 interesting and cool facts about lithium, which is the third element on the periodic table. Plus, common uses for the lightest metal.

Explore detailed information on Lithium (Li) element - Atomic Number 3, atomic mass 6.941, physical &

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chemical properties, facts, uses, history, periodic table trends, isotopes, electron configuration, ...

In the periodic table, lithium is the first alkali. Lithium is a rare element that is mostly found in molten rock and very small amounts in saltwater. It has the lowest melting point of any solid metal, ...

Lithium, chemical element of Group 1 (Ia) in the periodic table, the alkali metal group, lightest of the solid elements. The metal itself--which is soft, white, and lustrous--and several of its ...

Lithium is the third element of the periodic table. These lithium facts contain chemical and physical data along with general information and history.

Lithium is a Group 1 (IA) element containing just a single valence electron ($1s^2 2s^1$). Group 1 elements are called "alkali metals". Lithium is a solid only about half as dense as water and lithium metal is the ...

Lithium was discovered from a mineral, while other common alkali metals were ...

Lithium - Properties, history, name origin, facts, applications, isotopes, electronic configuration, crystal structure, hazards and more; Interactive periodic table of ...

Chemical element, Lithium, information from authoritative sources. Look up properties, history, uses, and more.

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